## Glutamate

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#### Abstract

Azomethine ylides have been generated from aromatic aldimines of $\alpha$-amino acid methyl esters under treatment with $\mathrm{LiBr} / \mathrm{Et}_{3} \mathrm{~N}$ and trapped with tert-butyl acrylate yielding racemic orthogonally protected cis-5-arylpyrrolidine-2,4-dicarboxylates regio- and stereo-selectively in high yields. Subsequent $N$-methylation and deprotection of 4-carboxylic group of cycloaddition products led to novel prolineglutamate $c i s$-chimeras with substituents at 2 and 5 positions of pyrrolidine scaffold.


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Introduction.
Combination of two amino acids fragments in one scaffold, typically conformationally constrained, is an actively used approach to non-proteinogenic $\alpha$-amino acids and subsequent construction of peptidomimetics. Pyrrolidine 2,4-dicarboxylic acid (1) contains structural features of proline and glutamic acid, and can be considered as a chimera of these two proteinogenic $\alpha$ amino acids. trans-Isomer of $\mathbf{1}$ was isolated from the seeds of Afzelia bella [1] and a potent competitive inhibition of glutamate transport was demonstrated for this compound [2]. Both cis- and trans-isomers of $\mathbf{1}$ have been used for synthesis of conformationally restricted glutathione analogues [3]. Bulky substituent in 5-position of proline ring influences on cis-/transpopulation of a peptide bond formed by nitrogen atom of proline [4]. From other side aryl substituent in 5position improves pharmacological properties of proline derivatives, namely mercaptoacyl 5-phenyl proline (2) was identified as 3 -fold more active than captopril in inhibition of angiotensin converting enzyme (ACE) [5]. The insertion of a phenyl group to the 5-position of proline in a series of 2-mercapto-3-phenylpropanoyl-Gly-(5-Ph)-Pro compounds led to discovery of very potent orally active dual inhibitor 3 of neutral endopeptidase (NEP) and ACE endowed with a long duration of action [6]. During our discovery program there were needed cis-pyrrolidine-2,4-dicarboxylates (4) with diverse aryl substituents in 5-position. Additional diversity points were designed at 2- and 4-positions of pyrrolidine ring by insertion of
$\mathrm{R}^{2}$-substituent and stepwise modification of orthogonally protected carboxylic functions. cis-5-Arylpyrrolidine-2,4-dicarboxylates (4) have been planned to use as building blocks for construction of appropriate libraries and subsequent high throughput screening [7]. This purpose prompted us to develop stereoselective, scalable and effective approach to pyrrolidine-glutamate chimeras 4 and the results of synthetic efforts towards the goal are presented here.


1



3



4

Figure 1. Structural formula of proline-glutamate chimeras and 5-arylprolines discussed in the introduction.

Table 1
Substituents in synthesized compounds.

| compound | $R^{l}$ | $R^{2}$ | compound | $R^{l}$ | $R^{2}$ |
| :---: | :--- | :--- | :---: | :--- | :--- |
|  |  |  |  |  |  |
| $\mathbf{a}$ | $2-\mathrm{F}$ | H | $\mathbf{g}$ | $2-\mathrm{Cl}$ | Me |
| $\mathbf{b}$ | $2,3-\mathrm{diF}$ | H | $\mathbf{h}$ | $4-\mathrm{Cl}$ | Me |
| $\mathbf{c}$ | $2-\mathrm{F}-5-\mathrm{MeO}$ | H | $\mathbf{i}$ | $4-\mathrm{MeO}$ | Et |
| $\mathbf{d}$ | $3-\mathrm{Cl}$ | H | $\mathbf{j}$ | $4-\mathrm{F}$ | Et |
| $\mathbf{e}$ | H | Me | $\mathbf{k}$ | 3-pyridyl | $i-\mathrm{Bu}$ |
| $\mathbf{f}$ | $2-\mathrm{Me}$ | Me |  |  |  |

## Results and Discussion.

The most satisfactory conditions for conducting the synthesis of compounds 4 were found to be those favored by Tsuge $[8,9]$ and Grigg [10,11]. Both groups extensively studied 1,3-cycloaddition of electron-deficient olefins to metallo-azomethine ylides generated from $\alpha$-amino acid esters Schiff bases.
majority of recent published methods of similar diversityoriented cycloadditions [13,14]. Lithium bromide is also typical reagent for metallo-1,3-dipole generation, but in many cases cycloaddition is accompanied with Michael addition products [10]. Nevertheless we have tried to apply lithium bromide on the basis of it lower cost for production of $N$-metalated azomethine ylides 6 and succeeded in isolation of orthogonally protected prolines 7 in $80-97 \%$ yield as single stereoisomers (Table 1, Scheme 2).
Stereochemistry of synthesized racemates 7 is a result of regio- and stereo-selective endo-cycloaddition process controlled by lithium chelation of both syn dipole 6 and tert-butyl acrylate (Scheme 2) and has been assigned by ${ }^{1} \mathrm{H}$ NMR spectroscopy. The cone anisotropy effect of the phenyl ring at C-5 causes a characteristic upfield shift on the protons of substituents at C-4 that are cis with regard to that phenyl group. For example, the proton signals of the

Scheme 1


Retrosynthesis of orthogonally protected $N$-methyl-5-arylpyrrolidine-2,4-dicarboxylates.

Since we aimed to have orthogonally protected carboxylic functions and methylated nitrogen atom [12] at the pyrrolidine scaffold 4 the retrosynthetic Scheme 1 was developed. It is based on cycloaddition of tert-butyl acrylate with 1,3-dipole generated from iminoester obtained in its turn from benzaldehyde and $\alpha$-amino acid methyl ester.

Required iminoesters 5 were readily obtained from substituted benzaldehydes and methyl esters of glycine, alanine, 2-aminobutyric acid, leucine under dehydrating conditions (Scheme 2). Silver (I) salts were used in the
tert-butoxycarbonyl group at C-4 in compounds 7 appear around 1.0 ppm . This value is 0.4 ppm more shielded than that of the tert-butoxycarbonyl groups protons in tert-butyl esters of aliphatic carboxylic acids (typically at 1.4 ppm ). This effect is also observed in the protons of the methyl group at C-4 of similar pyrrolidine-2,4-dicarboxylates [8,15]. It is worth to note that 1,3-dipolar cycloaddition occurs with sterically hindered imine $\mathbf{5 k}\left(\mathrm{R}^{2}=i-\mathrm{Bu}\right)$ under the general used conditions while the cycloaddition of iminoester with $\mathrm{R}^{2}=i-\operatorname{Pr}$ is reported to take place only under treatment with stronger base DBU [8].

Scheme 2


Reaction conditions: a) $\mathrm{MgSO}_{4}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt; b) t-Bu acrylate, $\mathrm{LiBr}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{THF}$, rt.

Two approaches were used for subsequent N -methylation and appropriate deprotection of 4-carboxylic function in pyrrolidine-2,4-dicarboxylates (7). The first is a two-step consequent alkylation - acidic cleavage of tert-butyl ester procedure (Scheme 3, steps $a$ and $b$ ). It works quite acceptable for simple 5-aryl substituents (2-F-phenyl (a), phenyl (e)) but in the case of 3-pyridyl substituent (compound $\mathbf{7 k}$ ) the quaternization of heteroaromatic nitrogen was the only observed process under $\mathrm{CH}_{3} \mathrm{I}$ treatment. This problem has been overcome with reductive methylation of pyrrolidine $\mathbf{7 k}$ using formaldehyde and formic acid (Scheme 3, step $c$ ) resulted in formation of nicotine analog $9 \mathbf{k}$. The same procedure was applied for compounds $\mathbf{7 b}-\mathbf{d}, \mathbf{f}-\mathbf{j}$ and N -methyl-pyrrolidine-2,4-dicarboxylates ( $\mathbf{9 b - d}, \mathbf{f}-\mathbf{j}$ ) were isolated in $35-65 \%$ yield that is comparable with $35-50 \%$ combined yield for two-step procedure.
establish stereochemistry of amino acids 9 X-ray analysis of dicarboxylate $9 \mathbf{b}$ has been done [16]. In the molecule $\mathbf{9 b}$, all bond lengths lie in the ordinary ranges for organic compounds. In crystal, two enantiomeric molecules related by symmetry centre are combined into hydrogenbonded dimeric units (figure 2).

In conclusion, an effective method of synthesis of $N$ -methyl-cis-5-arylpyrrolidine-2,4-dicarboxylates (9) based on 1,3-dipolar cycloaddition was developed. Different protection of carboxylic functionalities in chimeras 7 and 9 allows their subsequent independent transformations.

## EXPERIMENTAL

Melting points are uncorrected. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker AMX-400 spectrometer in dimethyl sulfoxide- $\mathrm{d}_{6}$ solutions. Chemical shifts ( $\delta, \mathrm{ppm}$ ) were referenced

Scheme 3


Reaction conditions: a) $\mathrm{CH}_{3} \mathrm{I}, \mathrm{DMF}$, rt; b) $\mathrm{HCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, rt; c) $\mathrm{HCO}_{2} \mathrm{H}, \mathrm{HCOH}$, reflux.

One of the marked features of compounds 9 is a big magnitude of ${ }^{3} \mathbf{J}_{(\mathrm{H} 4-\mathrm{H} 5)}$ coupling constant according ${ }^{1} \mathrm{H}$ NMR experiments - 10.0-10.5 Hz. To undoubtedly


Figure 2. Hydrogen - bonded dimers in the crystal structure of $\mathbf{9 b}$. $\mathrm{O}(1)-\mathrm{H}(1) 0.90(2), \mathrm{H}(1) \ldots \mathrm{O}(2 \mathrm{~A}) 1.76(2), \mathrm{O}(1) \ldots \mathrm{O}(2 \mathrm{~A}) 2.657(1)$ $\AA ; \mathrm{O}(1)-\mathrm{H}(1) \ldots \mathrm{O}(2 \mathrm{~A}) 177(2)^{\circ}$
to dimethyl sulfoxide $-\mathrm{d}_{5}$ signal $\left(\delta_{\mathrm{H}}=2.50 \mathrm{ppm}\right)$ and coupling constants (J) are indicated in Hz .

General Procedure for the Synthesis of Iminoesters 5 [14].
Under inert atmosphere corresponding benzaldehyde (1.0 eq) was added to a suspension of amino acid methyl ester hydrochloride (1.1 eq), $\mathrm{MgSO}_{4}(2.0 \mathrm{eq})$, and $\mathrm{NEt}_{3}$ ( 1.1 eq ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Resulted mixture was stirred at rt overnight. The solid was removed by filtration and the filtrate was washed once with $\mathrm{H}_{2} \mathrm{O}$. The aqueous phase was extracted once with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and the combined organic layers were washed with brine. The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated. Products were sufficiently pure by ${ }^{1} \mathrm{H}$ NMR and used in the next step without additional purification.

General Procedure for the Cycloaddition Reactions of tert-Butyl acrylate with Aldimines 5.

Solution of $4.18 \mathrm{~g}(48.0 \mathrm{mmol})$ of LiBr in dry THF ( 16 ml ) was added dropwise to the stirred mixture of 5.10 ml ( 35.2 mmol ) of tert-butyl acrylate, 32 mmol of aldimine 5 and 54 ml of dry THF under inert atmosphere. After cooling to $5^{\circ} \mathrm{C} 5.40 \mathrm{ml}$ ( 38.4 mmol ) of $\mathrm{Et}_{3} \mathrm{~N}$ were added dropwise with the rate not allowing the reaction temperature exceed $20^{\circ} \mathrm{C}$. The mixture was stirred at rt 24 h . The reaction was quenched by addition of a saturated aqueous solution of $\mathrm{NH}_{4} \mathrm{Cl}(60 \mathrm{ml})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 40 \mathrm{ml})$. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and the solvent was evaporated.

The crude products were purified by column chromatography on silica gel using hexane/AcOEt as an eluent.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Fluorophenyl)pyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7a).

This compound was obtained as a white solid, $\mathrm{mp} 53^{\circ} \mathrm{C}$. Yield $93 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.96\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 2.15-2.23(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3)$; 2.26-2.34 (m, 1H, H-3); 3.23 (dd, 1H, J 14.9, 7.1, H-4); 3.70 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 3.86 (dd, 1H, J 7.6, 7.6, H-2); 4.63 (m, 1H, H-5); 7.09-7.18 (m, 2H, Ar); 7.30 (dddd, 1H, J 13.5, 7.6, 5.4, 1.7, Ar); 7.51 (ddd, 1H, J 7.6, 1.7, 1.7, Ar).

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{FNO}_{4}$ : C, 63.14; $\mathrm{H}, 6.86 ; \mathrm{N}, 4.33$. Found: C, 62.95; H, 6.80; N, 4.24.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2,3-Difluorophenyl)pyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7b).

This compound was obtained as a white solid, $\mathrm{mp} 89^{\circ} \mathrm{C}$. Yield $92 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 1.00\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right.$ ); 2.16-2.33 (m, 2H, H-3); 3.25-3.31 (m, 1H, H-4); 3.69 (s, 3H, COOCH ${ }_{3}$ ); 3.82-3.91 (m, 1H, H-2); 4.71 (dd, J 7.6, 7.6, 1H, H-5); 7.13-7.20 (m, 1H, Ar); 7.26-7.33 (m, 1H, Ar); 7.33-7.39 (m, 1H, Ar).

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{21} \mathrm{~F}_{2} \mathrm{NO}_{4}$ : C, 59.82; H, 6.20; N, 4.10. Found: C, 59.59; H, 6.12; N, 4.20.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Fluoro-5-methoxyphenyl)pyrrolidine-2,4dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7c).

This compound was obtained as a white solid, $\mathrm{mp} 62-63^{\circ} \mathrm{C}$. Yield $80 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 1.00\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right.$ ); 2.14-2.31 (m, 2H, H-3); 3.20 (ddd, J 8.3, 8.3, 8.3, 1H, H-4); 3.69 (s 3H, ArOCH ${ }_{3}$ ); 3.71 (s, 3H, COOCH ${ }_{3}$ ); 3.85 (dd, J 8.5, 8.5, 1H, H-2); 4.63 (d, J 8.3, 1H, H-5); 6.81 (ddd, J 8.8, 3.3, 3.3, 1H, Ar); 7.00-7.11 (m, $2 \mathrm{H}, \mathrm{Ar})$.
Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{FNO}_{5}$ : C, $61.18 ; \mathrm{H}, 6.85 ; \mathrm{N}, 3.96$. Found: C, $60.90 ;$ H, 6.85 ; N, 3.89 .
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(3-Chlorophenyl)pyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7d).

This compound was obtained as an oil. Yield $95 \%$. ${ }^{1} \mathrm{H}$ NMR: ס $1.02\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 2.13-2.29(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3) ; 3.22$ (ddd, J 7.8, $7.8,8.6,1 \mathrm{H}, \mathrm{H}-4) ; 3.27$ (s, 1H, NH); 3.69 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 3.86 (dd, J 8.3, 8.3, 1H, H-2); 4.46 (d, J 8.6, 1H, H-5); 7.26-7.32 (m, 3H, Ar); 7.39-7.41 (m, 1H, Ar).
Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{ClNO}_{4}: \mathrm{C}, 60.09 ; \mathrm{H}, 6.53 ; \mathrm{N}, 4.12$. Found: C, 60.30; H, 6.65; N, 4.00.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-2-Methyl-5-phenylpyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7e).

This compound was obtained as a white solid, $\mathrm{mp} 50-52^{\circ} \mathrm{C}$. Yield $85 \% .{ }^{1} \mathrm{H}$ NMR: $\delta 0.96\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 1.38\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$; 1.97 (dd, 1H, J 13.1, 7.8, H-3); 2.47-2.55 (m, 1H, H-3); 3.303.37 (m, 1H, H-4); 3.71 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.59 (d, 1H, J 8.3, H5); 7.19-7.26 (m, 1H, Ar); 7.27-7.32 (m, 4H, Ar).

Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{4}$ : C, 67.69; H, 7.89; $\mathrm{N}, 4.39$. Found: C, 67.80; H, 7.95; N, 4.59.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-2-Methyl-5-o-tolylpyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7f).
This compound was obtained as a white solid, $\mathrm{mp} 115-116^{\circ} \mathrm{C}$. Yield $89 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.89$ (s, $9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}$ ); 1.39 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ); 1.97 (dd, J 13.0, 8.3, 1H, H-3); 2.34 (s, 3H, $\mathrm{ArCH}_{3}$ ); 2.51-2.55 (m; 1H, H-3); 2.99 (br s, 1H, NH); 3.35-3.41 (m, 1H, H-4); 3.71
(s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.75 (d, J 8.5, 1H, H-5); 7.06-7.15 (m, 3H, Ar); 7.33-7.39 (m, 1H, Ar).

Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{NO}_{4}$ : C, 68.44; H, 8.16; N, 4.20. Found: C, 68.19; H, 8.20; N, 4.39.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Chlorophenyl)-2-methylpyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7g).
This compound was obtained as a white solid, $\mathrm{mp} 102-104^{\circ} \mathrm{C}$. Yield 87\%. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.93$ (s, $9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}$ ); 1.38 (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ); 1.96 (dd, J 13.1, 8.3, 1H, H-3); 2.59 (dd; J 13.1, 6.1, 1H, H-3); 3.14 (br s, 1H, NH); 3.44 (ddd, J 8.3, 8.3, 6.1, 1H, H-4); 3.69 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.91 (d, J 8.3, 1H, H-5); 7.22-7.32 (m, 2H, Ar); 7.38 (dd, J 7.5, 1.8, 1H, Ar); 7.59 (dd, J 7.5, 1.8, 1H, Ar).

Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{ClNO}_{4}$ : C, 61.10; H, 6.84; N, 3.96. Found: C, 61.29; H, 6.90; N, 4.09.
$\left(2 S^{*}, 4 S^{*}, 5 R^{*}\right)$-5-(4-Chlorophenyl)-2-methylpyrrolidine-2,4dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7h).

This compound was obtained as a white solid, $\mathrm{mp} 110-111^{\circ} \mathrm{C}$. Yield $97 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.98$ ( $\mathrm{s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}$ ); $1.35\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$; 1.91 (dd, J 13.1, 7.8, 1H, H-3); 2.44-2.54 (m; 1H, H-3); 3.13 (br $\mathrm{s}, 1 \mathrm{H}, \mathrm{NH}) ; 3.25-3.36(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4) ; 3.68$ (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.60 (d, J 8.8, 1H, H-5); 7.27-7.38 (m, 4H, Ar).

Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{ClNO}_{4}$ : C, 61.10; H, 6.84; N, 3.96 . Found: C, 61.25; H, 6.80; N, 4.19.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-2-Ethyl-5-(4-methoxyphenyl)pyrrolidine-2,4dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7i).

This compound was obtained as an oil. Yield $97 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.79$ (t, J 7.3, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); 1.00 ( $\mathrm{s}, 9 \mathrm{H}, \mathrm{t}_{\mathrm{C}}^{4} \mathrm{H}_{9}$ ); 1.57-1.77 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); 1.98 (dd, J 13.5, 7.8, 1H, H-3); 2.43 (dd, J 13.5, 5.6, 1H, H-3); 2.95 (d, J 9.5, 1H, NH); 3.20 (ddd, J 7.8, $7.8,5.6,1 \mathrm{H}, \mathrm{H}-4) ; 3.71\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COOCH}_{3}\right) ; 3.72$ (s, 3 H , $\mathrm{ArOCH}_{3}$ ); 4.43 (dd, J 9.5, 7.8, 1H, H-5); 6.87 (d, J 8.6, 2H, Ar); 7.18 (d, J 8.6, 2H, Ar).

Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{NO}_{5}$ : C, 66.09; H, 8.04; N, 3.85. Found: C, 65.95; H, 8.00; N, 3.70.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-2-Ethyl-5-(4-fluorophenyl)pyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (7j).

This compound was obtained as a white solid, $\mathrm{mp} 39^{\circ} \mathrm{C}$. Yield $95 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.80\left(\mathrm{t}, \mathrm{J} 7.3,3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ); $1.00(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-$ $\mathrm{C}_{4} \mathrm{H}_{9}$ ); 1.59-1.78 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ); 1.99 (dd, J 13.5, $7.8,1 \mathrm{H}, \mathrm{H}-$ 3); 2.45 (dd, J 13.5, 5.7, 1H, H-3); 3.04 (br s, 1H, NH); 3.24 (ddd, J 7.8, 7.8, 5.7, 1H, H-4); 3.70 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.51 (m, 1H, H-5); 7.10-7.16 (m, 2H, Ar); 7.28-7.35 (m, 2H, Ar).
Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{26} \mathrm{FNO}_{4}$ : C, 64.94; H, 7.46; N, 3.99. Found: C, 65.05; H, 7.58; N, 3.79.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-2-Isobutyl-5-pyridin-3-yl-pyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester ( $7 \mathbf{k}$ ).

This compound was obtained as an oil. Yield $92 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.83\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.4, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 0.91(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.4$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 0.99\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 1.61-1.81(\mathrm{~m}, 3 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} ; 2.05$ (dd, 1H, J 13.2, 7.8, H-3); 2.45 (dd, 1H, J 13.2, 6.4, H-3); 3.36 (ddd, 1H, J 7.8, 7.8, 6.4, H-4); 3.73 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.60 (m, 1H, H-5); 7.37 (dd, 1H, J 7.8, 4.9, Ar); 7.69 (ddd, 1H, J 7.8, 1.7, 1.7, Ar); 8.46 (dd, 1H, J 4.9, 1.7, Ar); 8.52 (d, 1H, 1.7, Ar).

Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{4}$ : C, 66.27; $\mathrm{H}, 8.34 ; \mathrm{N}, 7.73$. Found: C, 66.18; H, 8.30; N, 7.90.

General Procedure for $N$-Methylation of Compounds 7 with $\mathrm{CH}_{3} \mathrm{I}$.
$\mathrm{CH}_{3} \mathrm{I}(0.38 \mathrm{ml}, 6.1 \mathrm{mmol})$ was added dropwise to the mixture of 4.1 mmol of $7,2.31 \mathrm{~g}(18.3 \mathrm{mmol})$ of $\mathrm{K}_{2} \mathrm{CO}_{3}$ and 17 ml of dry DMF. The mixture was stirred 24 h at rt. The solid was removed by filtration and the filtrate was concentrated in vacuum. Chromatography of the residue on silica gel using hexane/ AcOEt as an eluent afforded pure product $\mathbf{8}$.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Fluorophenyl)-1-methylpyrrolidine-2,4dicarboxylic acid 4-tert-butyl ester 2-methyl ester (8a).

This compound was obtained as an oil. Yield $94 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.93\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 2.13\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.18-2.37(\mathrm{~m}, 2 \mathrm{H}$, H-3); 3.20-3.35 (m, 2H, H-2, H-4); 3.70 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.09 (d, 1H, J 10.0, H-5); 7.07-7.22 (m, 2H, Ar); 7.24-7.35 (m, 1H, Ar); 7.47(ddd, 1H, J 7.6, 7.6, 1.0, Ar).
Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{FNO}_{4}$ : C, 64.08; H, 7.17; N, 4.15. Found: C, 63.96; H, 7.20; N, 4.30.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-1,2-Dimethyl-5-phenylpyrrolidine-2,4-dicarboxylic acid 4-tert-butyl ester 2-methyl ester (8e).
This compound was obtained as an oil. Yield $66 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.90\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{t}-\mathrm{C}_{4} \mathrm{H}_{9}\right) ; 1.25\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) ; 1.87(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J} 12.0$, 7.9, H-3); 2.03 (s, 3H, N-CH ); 2.59 (dd, 1H, J 12.0, 12.0, H-3); 3.40 (ddd, 1H, J 10.7, 10.7, 7.9, H-4); 3.69 (s, 3H, $\mathrm{COOCH}_{3}$ ); 3.95 (d, J 10.7, 1H, H-5); 7.15-7.34 (m, 5H, Ar).

Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{NO}_{4}$ : C, 68.44; $\mathrm{H}, 8.16 ; \mathrm{N}, 4.20$. Found: C, 68.66; H, 8.20; N, 4.37.

General Procedure for the Preparation of Compounds 9 from 8.
4.0 mmol of corresponding compound $\mathbf{8}$ were dissolved in 50 ml of dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Dry HCl was passed through the solution during 1 h at $5^{\circ} \mathrm{C}$. The mixture was stirred overnight at rt , washed with $\mathrm{H}_{2} \mathrm{O}(2 \times 10 \mathrm{ml}), 10 \%$ aqueous solution of $\mathrm{NaHCO}_{3}(2 \times 10$ $\mathrm{ml})$. Aqueous phases were extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{ml})$. The combined organic layers were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated. The residue was purified by column chromatography on silica gel using hexane/AcOEt as an eluent.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Fluorophenyl)-1-methylpyrrolidine-2,4dicarboxylic acid 2-methyl ester (9a).
This compound was obtained as a white solid, $\mathrm{mp} 126^{\circ} \mathrm{C}$. Yield 51\%. ${ }^{1} \mathrm{H}$ NMR: $\delta 2.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.23-2.36(\mathrm{~m}, 2 \mathrm{H}$, H-3); 3.24-3.35 (m, 2H, H-4, H-2); 3.70 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.11 (d, 1H, J 10.0, H-5); 7.09 (ddd, 1H, J 9.5, 8.3, 1.0, Ar); 7.15 (ddd, 1H, J 7.6, 7.6, 1.0, Ar); 7.23-7.30 (m, 1H, Ar); 7.50 (ddd, 1H, J 7.6, 7.6, 1.7, Ar); 11.80 (s, 1H, COOH).
Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{FNO}_{4}$ : C, 59.78 ; H, 5.73; N, 4.98. Found: C, 60.01; H, 5.69; N, 4.79.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-1,2-Dimethyl-5-phenylpyrrolidine-2,4-dicarboxylic acid 2 -methyl ester ( $\mathbf{9 e}$ ).

This compound was obtained as a white solid, $\mathrm{mp} 156^{\circ} \mathrm{C}$. Yield $50 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 1.27$ (s, 3H, CH3) ; 1.89 (dd, 1H, J 11.7, 7.8, H-3); 2.06 (s, 3H, N-CH ); 2.60 (dd, 1H, J 11.7, 11.7, H-3); 3.40 (ddd, $1 \mathrm{H}, \mathrm{J} 10.5,10.5,7.8, \mathrm{H}-4)$; 3.70 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 3.99 (d, 1H, J 10.5, H-5); 7.16-7.29 (m, 3H, Ar); 7.31-7.35 (m, $2 \mathrm{H}, \mathrm{Ar}) ; 11.72$ (s, 1H, COOH).
Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NO}_{4}$ : C, 64.97; H, 6.91; N, 5.05. Found: C, 64.78; H, 6.93; N, 5.25.

General One-step Procedure for the Preparation of Compounds 9 from 7.

Corresponding compound $7(36.0 \mathrm{mmol})$ was dissolved in 180 ml of $98 \%$ formic acid, 27 ml of $37 \%$ aqueous formaldehyde were added and reaction mixture was heated at $100^{\circ} \mathrm{C}$ during 3040 min . After cooling to rt 150 ml of water were added and the mixture was neutralized with $\mathrm{Na}_{2} \mathrm{CO}_{3}$. Resulting mixture was extracted with $\mathrm{CHCl}_{3}(5 \times 70 \mathrm{ml})$. Organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and the solvent was evaporated. The crude products were purified by column chromatography on silica gel using hexane/ AcOEt as an eluent.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2,3-Difluorophenyl)-1-methylpyrrolidine-2,4dicarboxylic acid 2-methyl ester ( $\mathbf{9 b}$ ).

This compound was obtained as a white solid, $\mathrm{mp} 140^{\circ} \mathrm{C}$. Yield $40 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 2.18\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.24-2.33(\mathrm{~m}, 2 \mathrm{H}$, H-3); 3.29-3.40 (m, 2H, H-4, H-2); 3.70 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.17 (d, J 10.0, 1H, H-5); 7.13-7.20 (m, 1H, Ar); 7.24-7.34 (m, 2H, $\mathrm{Ar}) ; 11.90(\mathrm{~s}, 1 \mathrm{H}, \mathrm{COOH})$.

Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~F}_{2} \mathrm{NO}_{4}$ : C, $56.19 ; \mathrm{H}, 5.05 ; \mathrm{N}, 4.68$. Found: C, 56.35; H, 5.10; N, 4.89.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Fluoro-5-methoxyphenyl)-1-methylpyrrol-idine-2,4-dicarboxylic acid 2-methyl ester (9c).

This compound was obtained as a white solid, $\mathrm{mp} 95^{\circ} \mathrm{C}$. Yield $46 \%{ }^{1}{ }^{1} \mathrm{H}$ NMR: $\delta 2.15$ (s, 3H, N-CH ${ }^{2}$ ); 2.27 (td, 2H, J 8.3, 4.0, $\mathrm{H}-3$ ); 3.23-3.31 (m, 2H, H-4, H-2); 3.69 (s, $6 \mathrm{H}, \mathrm{COOCH}_{3}$, $\mathrm{ArOCH}_{3}$ ); 4.06 (d, J 9.9, 1H, H-5); 6.79 (dt; J 8.8, 3.7, 1H, Ar); 6.98-7.05 ( $2 \mathrm{H}, \mathrm{m} ; \mathrm{Ar}$ ); 11.85 (br s, $1 \mathrm{H}, \mathrm{COOH}$ ).

Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{FNO}_{5}$ : C, $57.87 ; \mathrm{H}, 5.83 ; \mathrm{N}, 4.50$. Found: C, 57.99; H, 5.75; N, 4.71.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(3-Chlorophenyl)-1-methylpyrrolidine-2,4dicarboxylic acid 2-methyl ester ( $\mathbf{9 d}$ ).

This compound was obtained as a white solid, $\mathrm{mp} 120-122^{\circ} \mathrm{C}$. Yield 57\%. ${ }^{1} \mathrm{H}$ NMR: $\delta 2.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.17-2.34(\mathrm{~m}, 2 \mathrm{H}$, H-3); 3.23-3.33 (m, 2H, H-4, H-2); 3.70 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 3.85 (d, J 10.0, 1H, H-5); 7.26-7.33 (m, 3H, Ar); 7.38 (dd, J 1.7, 1.7, $1 \mathrm{H}, \mathrm{Ar})$.

Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{ClNO}_{4}$ : C, 56.48; $\mathrm{H}, 5.42 ; \mathrm{N}, 4.70$. Found: C, 56.54; H, 5.46; N, 4.81.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-1,2-Dimethyl-5-o-tolylpyrrolidine-2,4-dicarboxylic acid 2-methyl ester (9f).

This compound was obtained as a white solid, $\mathrm{mp} 121-122^{\circ} \mathrm{C}$. Yield 50\%. ${ }^{1} \mathrm{H}$ NMR: $\delta 1.28$ (s, 3H, $\mathrm{CH}_{3}$ ); 1.93 (dd, J 12.0, 8.1, $1 \mathrm{H}, \mathrm{H}-3$ ); 2.02 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}$ ); 2.33 (s, 3H, $\mathrm{ArCH}_{3}$ ); 2.61 (dd, J 12.0, 10.3, 1H, H-3); 3.43 (ddd, J 10.3, 10.3, 8.1, 1H, H-4); 3.70 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.22 (d, J 10.3, 1H, H-5); 7.03-7.14 (m, 3H, Ar); 7.50 (d, J 7.3, 1H, Ar).

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{4}$ : C, 65.96; $\mathrm{H}, 7.27 ; \mathrm{N}, 4.81$. Found: C, 65.70; H, 7.20; N, 5.03.
( $2 S^{*}, 4 S^{*}, 5 R^{*}$ )-5-(2-Chlorophenyl)-1,2-dimethylpyrrolidine-2,4dicarboxylic acid 2-methyl ester ( $\mathbf{9 g}$ ).

This compound was obtained as a white solid, $\mathrm{mp} 130-131^{\circ} \mathrm{C}$. Yield $35 \%$. ${ }^{1}$ H NMR: $\delta 1.30\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right) ; 1.97$ (dd, J 12.2, 8.1, $1 \mathrm{H}, \mathrm{H}-3$ ); 2.06 (s, $3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}$ ); 2.61 (dd, J 12.2, 9.8, $1 \mathrm{H}, \mathrm{H}-3$ ); 3.45 (td, J 10.0, 8.1, 1H, H-4); 3.70 (s, 3H, $\mathrm{COOCH}_{3}$ ); 4.43 (d, J $10.0,1 \mathrm{H}, \mathrm{H}-5) ; 7.23$ (ddd, J 7.6, 1.7, 1.7, 1H, Ar); 7.30 (ddd, J
$7.6,1.5,1.5,1 \mathrm{H}, \mathrm{Ar}) ; 7.34$ (dd, J 7.6, 1.5, 1H, Ar); 7.58 (dd, J 7.6, 1.7, 1H, Ar).

Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClNO}_{4}: \mathrm{C}, 57.79 ; \mathrm{H}, 5.82 ; \mathrm{N}, 4.49$. Found: C, 57.85; H, 5.80; N, 4.30.
$\left(2 S^{*}, 4 S^{*}, 5 R^{*}\right)$-5-(4-Chlorophenyl)-1,2-dimethylpyrrolidine-2,4dicarboxylic acid 2-methyl ester (9h).

This compound was obtained as a white solid, mp $135-137^{\circ} \mathrm{C}$. Yield 50\%. ${ }^{1} \mathrm{H}$ NMR: $\delta 1.27$ (s, 3H, $\mathrm{CH}_{3}$ ); 1.90 (dd, J 11.7, 7.8, $1 \mathrm{H}, \mathrm{H}-3$ ); 2.06 (s, 3H, N-CH $\mathrm{N}_{3}$ ) 2.56 (dd, J 11.7, 11.7, 1H, H-3); 3.41 (ddd, J 10.5, 10.5, 7.8, 1H, H-4); 3.70 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 4.02 (d, J 10.5, 1H, H-5); 7.30-7.37 (m, 4H, Ar); 11.81 (br s, 1H, COOH ).

Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{ClNO}_{4}$ : C, $57.79 ; \mathrm{H}, 5.82 ; \mathrm{N}, 4.49$. Found: C, 57.77; H, 5.70; N, 4.32.
$\left(2 S^{*}, 4 S^{*}, 5 R^{*}\right)$-2-Ethyl-5-(4-methoxyphenyl)-1-
methylpyrrolidine-2,4-dicarboxylic acid 2-methyl ester (9i).
This compound was obtained as a white solid, mp $94-95^{\circ} \mathrm{C}$. Yield $36 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.84\left(\mathrm{t}, \mathrm{J} 7.3,3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; 1.53$ (ddd, J $7.3,7.3,7.3,1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); 1.92 (ddd J $7.3,7.3,7.3,1 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ) ; 2.00 (dd, J 12.5, 7.8, 1H, H-3); 2.17 (s, $3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}$ ); 2.67 (dd, J 12.5, 12.5, 1H, H-3); 3.24-3.33 (m, 1H, H-4); 3.69 (s, $\left.3 \mathrm{H}, \mathrm{COOCH}_{3}\right) ; 3.71\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{ArOCH}_{3}\right) ; 4.01(\mathrm{~d}, \mathrm{~J} 10.3,1 \mathrm{H}, \mathrm{H}-5)$; 6.81 (d, J 8.8, 2H, Ar); 7.18 (d, J 8.8, 2H, Ar); 11.69 (br s, 1H, $\mathrm{COOH})$.

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{NO}_{5}$ : C, 63.54; $\mathrm{H}, 7.21 ; \mathrm{N}, 4.36$. Found: C, 63.66; H, 7.20; N, 4.57.
$\left(2 S^{*}, 4 S^{*}, 5 R^{*}\right)$-2-Ethyl-5-(4-fluorophenyl)-1-methylpyrrolidine-2,4-dicarboxylic acid 2-methyl ester ( $\mathbf{( 9 j}$ ).

This compound was obtained as a white solid, mp $123-125^{\circ} \mathrm{C}$. Yield $67 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.85\left(\mathrm{t}, \mathrm{J} 7.3,3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ); 1.50-1.60 (m, 1H, CH2 CH 3 ); 1.88-1.98 (m, 1H, CH2CH3); 2.02 (dd, J 13.0, $7.8,1 \mathrm{H}, \mathrm{H}-3) ; 2.18\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.66$ (dd, J 13.0, 10.8, 1H, $\mathrm{H}-3$ ); 3.33 (ddd, J 10.8, 10.3, $7.8,1 \mathrm{H}, \mathrm{H}-4$ ); 3.69 (s, 3 H , $\mathrm{COOCH}_{3}$ ) ; 4.09 (d, J 10.3, 1H, H-5); 7.04-7.10 (m, 2H, Ar); 7.29 (ddd, J 12.0, 5.4, 3.1, 2H, Ar); 11.78 (s, 1H, COOH).

Anal. Calcd for $\mathrm{C}_{16} \mathrm{H}_{20} \mathrm{FNO}_{4}: \mathrm{C}, 62.13 ; \mathrm{H}, 6.52 ; \mathrm{N}, 4.53$. Found: C, 62.31; H, 6.38; N, 4.54.
$\left(2 S^{*}, 4 S^{*}, 5 R^{*}\right)$-2-Isobutyl-1-methyl-5-pyridin-3-yl-pyrrolidine-2,4-dicarboxylic acid 2-methyl ester ( $\mathbf{9 k}$ ).

This compound was obtained as a white solid, mp $156^{\circ} \mathrm{C}$. Yield $37 \%$. ${ }^{1} \mathrm{H}$ NMR: $\delta 0.81\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{J} 6.6, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 0.94$ (d, 3H, J 6.6, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 1.43(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J} 13.0,5.1$, $\left.\mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 1.53-1.62\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 1.78-1.87$ $\left(\mathrm{m}, 2 \mathrm{H}, \mathrm{H}-3, \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right) ; 2.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{N}-\mathrm{CH}_{3}\right) ; 2.73(\mathrm{dd}, 1 \mathrm{H}$, J 12.7, 12.7, H-3); 3.04 (ddd, 1H, J 12.7, 10.5, 7.1, H-4); 3.66 (s, $3 \mathrm{H}, \mathrm{COOCH}_{3}$ ); 3.84 (d, 1H, J 10.5, H-5); 7.14 (dd, 1H, J 7.8, 4.7, Ar); 7.60 (ddd, 1H, J 7.8, 1.7, 1.7, Ar); 8.27 (dd, 1H, J 4.7, 1.7, Ar); 8.38 (d, 1H, J 1.7, Ar).

Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{4}: \mathrm{C}, 63.73 ; \mathrm{H}, 7.55 ; \mathrm{N}, 8.74$. Found: C, 63.61; H, 7.48; N, 8.84.

## Crystal Structure Determination of Compound $\mathbf{9 b}$.

The single crystal of $\mathbf{9 b}$ of approximate dimensions 0.30 x $0.20 \times 0.10 \mathrm{~mm}$ was mounted in inert oil on the top of glass fibre and transferred to a cold nitrogen stream on the Bruker SMART

CCD diffractometer. Crystal data: $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~F}_{2} \mathrm{~N}_{1} \mathrm{O}_{4}, \mathrm{M}=299.27$, monoclinic, $a=10.2002(7), b=10.9433(7), c=12.4771(8) \AA$, $\beta=98.236(1)^{\circ}, V=1378.38(16) \AA^{3}$, space group $P 2_{1} / n, Z=4$, $D_{\text {c }}=1.442 \mathrm{~g} / \mathrm{cm}^{3}, F(000)=624, \mu(\mathrm{Mo}-\mathrm{K} \alpha)=0.123 \mathrm{~mm}^{-1}$. Total of 8677 reflections ( 3318 unique, $R_{\text {int }}=0.0247$ ) were measured using graphite monochromatized $\mathrm{Mo}-\mathrm{K} \alpha$ radiation $(\lambda=0.71073$ $\AA$ ) at $120.0(2) \mathrm{K}$. Data were collected in the range $2.42<\theta<$ $28.00(-8 \leq \mathrm{h} \leq 13,-14 \leq \mathrm{k} \leq 14,-16 \leq 1 \leq 14)$. Omega scan mode with the step of 0.3 deg ( 15 sec . per step) was used. The structure was solved by direct methods [17] and refined by full matrix least-squares on $F^{2}$ [18] with anisotropic thermal parameters for all non-hydrogen atoms. All H atoms were found from difference Fourier syntheses and refined in an isotropic approximation. The final residuals were: $R_{1}=0.0366, \mathrm{w} R_{2}=$ 0.0867 for 2621 reflections with $I>2 \sigma(I)$ and $0.0530,0.0926$ for all data and 250 parameters. Goof $=1.036$, maximum $\Delta \rho=$ $0.314 \mathrm{e} / \AA^{3}$.

## REFERENCES AND NOTES

[1] A. Welter, M. Marliert, and G. Dardenne, Phytochemistry, 17, 131 (1978).
[2] R. J. Bridges, M. S. Stanley, M. W. Anderson, C. W. Cotman, and A. R. Chamberlin, J. Med. Chem., 34, 717 (1991).
[3] M. P. Paradisi, A. Mollica, I. Cacciatore, A. Di Stefano, F. Pinnen, A. M. Caccuri, G. Ricci, S. Dupre, A. Spirito, and G. Lucente, Bioorg. Med. Chem., 11, 1677 (2003).
[4] E. A. A. Wallen, J. A. M. Christiaans, J. Gynther, and J. Vepsalainen, Tetrahedron Lett., 44, 2081 (2003).
[5] M. M. Murphy, J. R. Schullek, E. M. Gordon, and M. A. Gallop, J. Am. Chem. Soc., 117, 7029 (1995).
[6] M. Fournie-Zaluski, P. Coric, V. Thery, W. Gonzalez, H. Meudal, S. Turcaud, J. Michel, and B. P. Roques, J. Med. Chem., 39, 2594 (1996).
[7] The libraries construction and screening experiments are in progress. These results will be presented elsewhere.
[8] O. Tsuge, S. Kanemasa, and M. Yoshioka, J. Org. Chem., 53, 1384 (1988).
[9] O. Tsuge, and S. Kanemasa, in Advances in Heterocyclic Chemistry, Vol 45, A. R. Katritzky, ed, Academic Press, SanDiego, 1989, pp 231-349.
[10] D. A. Barr, R. Grigg, H. Q. N. Gunaratne, J. Kemp, P. McMeekin, and V. Sridharan, Tetrahedron, 44, 557 (1988).
[11] R. Grigg, Chem. Soc. Rev., 16, 89 (1987).
[12] L. Aurelio, R. T. C. Brownlee, and A. B. Hughes, Chem. Rev., 104, 5823 (2004).
[13] M. Nyerges, D. Bendell, A. Arany, D. E. Hibbs, S. J. Coles, M. B. Hursthouse, P. W. Groundwater, and O. Meth-Cohn, Synlett, 947 (2003).
[14] C. Chen, X. Li, and S. L. Schreiber, J. Am. Chem. Soc., 125, 10174 (2003).
[15] P. B. Woller, and N. H. Cromwell, J. Org. Chem., 35, 888, (1970); S. Cabrera, R. G. Arrayas, and J. C. Carretero, J. Am. Chem. Soc., 127, 16394 (2005).
[16] Crystallographic data (excluding structure factors) for the structure reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC604160. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [Fax: int.code + 44(1223)336-033; E-mail: deposit@ccdc.cam.ac.uk].
[17] G. M. Sheldrick, Acta Cryst., A46, 467 (1990).
[18] G. M. Sheldrick, SHELXL-97, Program for the Refinement of Crystal Structures. University of Gottingen, 1997.

